#### Last updated 7/1/23

These concepts have been greatly simplified

Crystals are characterized with 3d geometric structures (unit cell)



The spheres here represent the approximate extents of the atom, not the nuclei

- Silicon unit cell
  - 2 interpenetrating Face Centered Cubic Cells



2 FCC cells 1 shifted ¼ of the way along the major diagonal



Resulting diamond cubic lattice



Atoms showing 4 nearest neighbor structure

# Lattice parameter (edge length) of 0.543 nm Nearest neighbor distance is 0.235 nm.

- Silicon unit cell
  - Corners: 8, 1/8 atoms/unit cell
  - Faces: 6, 1/2 atoms/unit cell
  - Inside: 4 full atoms/unit cell

3 <u>4</u> 8 atoms/unit cell

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Cell volume:

 $(.543 \text{ nm})^3 = 1.6 \times 10^{-22} \text{ cm}^3$ 

Density of silicon atoms

= (8 atoms) / (cell volume) = 5 x 10<sup>22</sup> atoms/cm<sup>3</sup>

- Silicon unit cell bonding model
  - Si has a half full outer electron shell (4 / 8)
  - Lowest energy state is the Diamond Lattice
    - 4 electrons are shared between 4 nearest neighbors



Atoms showing 4 nearest neighbor structure



Each atom appears to have a full valence band  $\rightarrow$  very stable