

Corrected (10/16/08)

$$5. P_{\text{initial}} = P^{\text{sat}}(50^\circ\text{C}) = \underline{\underline{12.352 \text{ kPa}}}$$

$$\text{Final state } P = 100 \text{ kPa}, v = 3.01 \text{ m}^3/\text{kg} \quad 5 \text{ PTS}$$

(no change)

$v > v_{\text{sat}} @ 100 \text{ kPa}$ so superheated.

| | <u>T (°C)</u> | <u>v (m³/kg)</u> | <u>u (kJ/kg)</u> |
|---|---------------|-----------------------------|------------------|
| 1 | 300 | 2.6389 | 2810.7 |
| | . | 3.01 | 2937.5 |
| 2 | 400 | 3.1027 | 2968.3 |

$$T = 300 + 100 \frac{3.01 - 2.64}{3.10 - 2.64} = \underline{\underline{380^\circ\text{C}}} \quad 10 \text{ PTS}$$

$$V_{\text{tank}} = (0.500 \text{ kg})(3.01 \text{ m}^3/\text{kg}) = \underline{\underline{1.50 \text{ m}^3}} \quad 5 \text{ PTS}$$

$$u = 2810.7 + 1576 \left(\frac{3.01 - 2.64}{3.10 - 2.64} \right) = \underline{\underline{2937.5}}$$

Rigid tank so $Q = \Delta U = m \Delta u$

$$u_i = 0.75 (209.33) + 0.25 (2442.7) \text{ kJ/kg}$$
$$= 767.67 \text{ kJ/kg}$$

$$Q = 0.500 \text{ kg} (2937.5 - 767.67 \text{ kJ/kg})$$
$$= \underline{\underline{1085 \text{ kJ}}}$$

using $Q = AH$ was -3 (answer was 1337 kJ)