## Exam 1 v.1.0 Key BE-382, Winter '08-'09, Dr. C. S. Tritt

This is a 1 hour closed book, closed notes exam. Write your answers on the paper provided or your own and organize and explain you work for full and partial credit. You may use 1 double sided, 8½ by 11 inch equation sheet, EES and the provided Units Conversion Factors and Properties handout. If you use EES, e-mail me your files (the file names should encode your last name and the problem number) at the end of the exam and clearly indicate on your exam that you used EES. Please turn in your units conversion handout at the end of the test. Put your name on it if you make any marks or notes in it. There are 3 problems, point values are as indicated.

Conversion factors and the Moody chart are attached.

 (30) The viscosity of a sample of unknown fluid is to be measured in a concentric cylinder device at 37°C. The radius of the inner cylinder is 8.00 cm, the gap between the cylinders is 1.00 mm and the length of the cylinders is 10.0 cm. The speed of the inner cylinder is varied and the following torques are measured:

Speed (rpm)	Torque (N·m)
300	20.21
450	30.32
600	40.43

What is the viscosity or apparent viscosities of the fluid and is it Newtonian under the test conditions?

- 2. (35) It has been proposed that a non-lethal (or at least unlike to be deadly) device be developed to protect ships from pirates in the Gulf of Aden. In particular, a water jet device is being developed. Assuming uniform cross-sectional velocity, what velocity (in ft/s) would a 2.00 inch diameter water ( $\rho = 62.3 \text{ lb}_m/\text{ft}^3$ ) jet require to produce a 200 lb<sub>f</sub> force on its target. Assume the water spatters uniformly and particularly upon impact.
- (35) In an operating room, cooling water flows through a 1.00 cm ID tube at a rate of 6.0 l/min. The tube is 20.0 m long. Given the density of the water is 1002 kg/m<sup>3</sup> and its viscosity is 1.00 cp, what is the pressure drop, in Pa, from end-to-end of the tube?

1. Viscosity determination - concentric cylinder  
General equation 9-12: device.  

$$T = \mathcal{U} = \frac{4\pi^2 R^3 \dot{n} L}{l} \Rightarrow \mathcal{U} = \frac{lT}{4\pi^2 R^3 \dot{n} L}$$

$$R = 8.00 \text{ cm} = 0.080 \text{ m}$$

$$L = 20.0 \text{ cm} = 0.200 \text{ m}$$

$$\dot{n} = 450 \text{ rpm} = \frac{450 \text{ /min}}{60 \text{ s/min}} = 7.5 \text{ s}^{-1}$$

$$l = 0.10 \text{ cm} = 0.00100 \text{ m}$$
2nd case:

$$M = \frac{(0.00100 \text{ m})(30.32 \text{ N} \cdot \text{m})}{4 \pi^2 (0.080^3 \text{ m}^3)(7.5 \text{ s}^{-1})(0.100 \text{ m})}$$
  
=  $1.00 \frac{\text{N} \cdot \text{s}}{\text{m}^2}$   
1<sup>st</sup> ruse :  $M = 2.00 \frac{\text{N} \cdot \text{s}}{\text{m}^2}$   
3<sup>rd</sup> case :  $M = 2.00 \frac{\text{N} \cdot \text{s}}{\text{m}^2}$   
So fluid is New Toniah.

2. Water jet impact force.  
Start with equation from Example 13-4.  
(like equation 13-23).  

$$F = \beta \ \dot{m} V_1$$
 10 pts  
Uniform cross-sectional velocity  $\Rightarrow \beta = 1$   
 $F = \dot{m} V_1$  3pts  
Volumetric flow,  $\dot{V} = V A_{cs}$  and  
mass flow,  $\dot{m} = \dot{V} \ell$ , so the  
equation becomes  
 $F = \ell A_{cs} V^2$  or  $V = \sqrt{\frac{F}{\ell} A_{cs}}$   
 $F = 200 \ lbf$   
 $A_{cs} = \pi r^2 = \pi (\frac{1}{12} fr)^2 = 0.0218 \ frz$   
 $\ell = 62.3 \ \frac{lbm}{A^3} (given)$   
 $V = \sqrt{\frac{200 \ lbf}{(62.3 \ lbm}} (0.0218 \ frz)}$   
 $= \sqrt{4740 \ fry_{sz}} = 68.8 \ fry_{sz}$   
Note this velocity is about 47 mph  
so it seems reasonable.  
Not converting units  $-7$ 

3: Pressure drop in a ppe flow.  

$$\frac{L}{D} = \frac{20 \text{ m}}{0.01 \text{ m}} = 2,000 \text{ so assume} \\ \text{fully developed.} \\ \text{Plan to use friction factor:} \\ AP = \int \frac{L}{D} \frac{e V_{ave}^2}{2} \quad (E_0, 14-21) \\ \text{Vave} = \frac{V}{Acs} = \frac{1.00 \times 10^{-4} \text{ m}_3^2}{7.95 \times 10^{-5} \text{ m}^2} = 1.27 \text{ m/s} \\ V = 6.00 \frac{P_{min}}{1000 \text{ e}} \left(\frac{1 \text{ m}^3}{60 \text{ s}}\right) = \frac{1.00 \times 10^{-4} \text{ m}_3}{1.00 \times 10^{-4} \text{ m}_3} \\ \text{Acs} = \pi r^2 = \frac{\pi D^2}{4} = \frac{\pi (D.01 \text{ m})^2}{4} = 7.85 \times 10^{-5} \text{ m}^2} \\ \text{N}_{Re} = \frac{e V_{ave} D}{M} = \frac{(1002 \frac{V_{max}}{M})(1.27 \frac{V}{8})(0.01 \text{ m})}{1.00 \times 10^{-3} \frac{V}{9}} \\ \text{M} = 1.00 \text{ cp} \left(\frac{1 \text{ porse}}{100 \text{ cf}}\right) \left(\frac{1 \text{ K}}{10 \text{ pase}}\right) = 1.00 \times 10^{-3} \frac{K_0}{m.s} \\ \text{Note; Viscosity is defined and discussed} \\ \text{N}_{Pe} = 12,700, \text{ so flow is turbulent} \end{cases}$$

Assume a Smooth pipe 
$$(\frac{e}{D} = 0)$$
 or  
smooth rubber  $(\frac{e}{D} = 0.01 \text{ mm}/10 = 0.001)$ .  
From chart  
 $f = 0.029 (\text{smooth})$  or  
 $f = 0.031 (\text{rubber})$   
 $AP = (0.029)(2000) \frac{(1002 \text{ Kg/ms})(1.27 \text{ m/s})^2}{2}$   
 $= \frac{4}{1.69 \times 10^4} \frac{\text{Kg}}{(\text{m.s})} \frac{\text{KPa}}{(1000 \text{ Kg/ms})}$   
 $= \frac{46.9 \text{ KPa}}{(1000 \text{ Kg/ms})} (\text{smooth} \frac{1}{2} \text{ arm}, \text{ which})$   
 $= \frac{46.9 \text{ KPa}}{(\text{or about } \frac{1}{2} \text{ arm}, \text{ which})}$   
The value for the rubber pipe is  
 $AP = \frac{50.1 \text{ KPa}}{2}$   
The Colebrook, Haaland or Swamee - Jaib  
equations could also be used.  
Wring  $\frac{5}{2} - \frac{5}{2}$  wrong  $\Delta P = \frac{1000}{2} \text{ arm} - \frac{100}{2}$   
wrong  $\Delta P = \frac{5}{2} (\frac{1000}{2} \text{ mooth} \frac{1000}{2} \text{ mooth}$ 

A couple of students appear to have confused Eq. 14-21 (which is for pressure loss) with Eq. 14-24 (which is for head loss). Both of these equations do, in fact relate to the energy loss in a flowing fluid, but the units are different (specifically by  $\rho q$ ).

## **Conversion Factors**

Acceleration	$1 \text{ m/s}^2 = 100 \text{ cm/s}^2$	$1 m/s^2 = 3.2808 ft/s^2$ $1 ft/s^2 = 0.3048* m/s^2$
Area	$1 \text{ m}^2 = 10^4 \text{ cm}^2 = 10^6 \text{ mm}^2$ = $10^{-6} \text{ km}^2$	$1 m^{2} = 1550 in^{2} = 10.764 ft^{2}$ $1 n^{2} = 144 in^{2} = 0.09290304* m^{2}$
Density	$f g/cm^3 = 1 kg/L = 1000 kg/m^3$	$1 \text{ g/cm}^3 = 62.428 \text{ lbm/ft}^3 = 0.036127 \text{ lbm/in}^3$ $1 \text{ lbm/in}^3 = 1728 \text{ lbm/ft}^3$ $1 \text{ kg/m}^3 = 0.062428 \text{ lbm/ft}^3$
Energy, heat, work, internal energy, enthalpy	$1 kJ = 1000 J = 1000 Nm = 1 kPa \cdot m^{3}$ $1 kJ/kg = 1000 m^{2}/s^{2}$ 1 kWh = 3600 kJ $1 cal^{*} = 4.184 J$ $I T cal^{*} = 4.1868 J$ $I Cal^{*} = 4.1868 kJ$	I kJ = 0.94782 Btu I Btu = 1.055056 kJ = 5.40395 psia $\cdot$ ft <sup>3</sup> = 778.169 lbf $\cdot$ ft I Btu/lbm = 25.037 ft <sup>2</sup> /s <sup>2</sup> = 2.326* kJ/kg I kJ/kg = 0.430 Btu/lbm I kWh = 3412.14 Btu I therm = 10 <sup>5</sup> Btu = 1.055 × 10 <sup>5</sup> kJ (natural gas)
Force	$1 N = 1 kg \cdot m/s^2 = 10^5 dyne$ 1 kgf = 9.80665 N	1 lbf = $32.174$ lbm $\cdot$ ft/s <sup>2</sup> = $4.44822$ N 1 N = $0.22481$ lbf
Heat flux	$1 \text{ W/cm}^2 = 10^4 \text{ W/m}^2$	$1 \text{ W/m}^2 = 0.3171 \text{ Btu/h} \cdot \text{ft}^2$
Heat generation rate	$1 \text{ W/cm}^3 \approx 10^6 \text{ W/m}^3$	$1 \text{ W/m}^3 = 0.09665 \text{ Btu/h} \cdot \text{ft}^3$
Heat transfer coefficient	$1 \mathbf{W/m^2} \cdot \mathbf{^oC} = 1 \mathbf{W/m^2} \cdot \mathbf{K}$	$1 \text{ W/m}^2 \cdot {}^{\circ}\text{C} = 0.17612 \text{ Btu/h} \cdot \text{ft}^2 \cdot {}^{\circ}\text{F}$
Length	1 m = 100 cm = 1000 mm 1 km = 1000 m	1 m = 39.370 in = 3.2808 ft = 1.0926 yd 1 ft = 12 in = 0.3048 * m 1 mile = 5280 ft = 1.6093 km 1 in = 2.54 * cm
Mass	1  kg = 1000  g 1  metric ton = 1000  kg	1 kg = $2.2046226$ lbm 1 lbm = $0.45359237*$ kg 1 ounce = $28.3495$ g 1 slug = $32.174$ lbm = $14.5939$ kg 1 short ton = $2000$ lbm = $907.1847$ kg

\*Exact conversion factor between metric and English units.

<sup>1</sup>Calorie is originally defined as the amount of heat needed to raise the temperature of 1 g of water by 1°C, but it varies with temperature. The international steam table (IT) calorie (generally preferred by engineers) is exactly 4.1868 J by definition and corresponds to the specific heat of water at 15°C. The thermochemical calorie (generally preferred by physicists) is exactly 4.184 J by definition and corresponds to the specific heat of water at room temperature. The difference between the two is about 0.06 percent, which is negligible. The capitalized Calorie used by nutritionists is actually a kilocalorie (1000 IT calories).

Power, heat transfer rate	1 W = 1 J/s 1 kW = 1000 W = 1.341 hp $1 hp^{\pm} = 745.7 W$	1 kW = 3412.14 Btu/h = 737.56 lbf · ft/s 1 hp = 550 lbf · ft/s = 0.7068 Btu/s = 42.41 Btu/min = 2544.5 Btu/h = 0.74570 kW 1 boiler hp = 33,475 Btu/h 1 Btu/h = 1.055056 kJ/h 1 ton of refrigeration = 200 Btu/min
Pressure	$1 Pa = 1 N/m^{2}$ $1 kPa = 10^{3} Pa = 10^{-3} MPa$ $1 atm = 101.325 kPa = 1.01325 bars$ $= 760 mmHg at 0^{\circ}C$ $= 1.03323 kgf/cm^{2}$ $1 mmHg = 0.1333 kPa$	1 Pa = $1.4504 \times 10^{-4}$ psia = 0.020886 lbf/ft <sup>2</sup> 1 psia = 144 lbf/ft <sup>2</sup> = 6.894757 kPa 1 atm = 14.696 psia = 29.92 inHg at 30°F t inHg = 3.387 kPa
Specific heat	$1 \text{ kJ/kg} \cdot {}^{\circ}\text{C} = 1 \text{ kJ/kg} \cdot \text{K}$ $= 1 \text{ J/g} \cdot {}^{\circ}\text{C}$	$1 \text{ Btu/lbm} \cdot {}^{\circ}\text{F} = 4.1868 \text{ kJ/kg} \cdot {}^{\circ}\text{C}$ $1 \text{ Btu/lbmol} \cdot \text{R} = 4.1868 \text{ kJ/kmol} \cdot \text{K}$ $1 \text{ kJ/kg} \cdot {}^{\circ}\text{C} = 0.23885 \text{ Btu/lbm} \cdot {}^{\circ}\text{F}$ $= 0.23885 \text{ Btu/lbm} \cdot \text{R}$
Specific volume	$1 \text{ m}^3/\text{kg} = 1000 \text{ L/kg}$ = 1000 cm <sup>3</sup> /g	$1 \text{ m}^3/\text{kg} = 16.02 \text{ ft}^3/\text{lbm}$ $1 \text{ ft}^3/\text{lbm} = 0.062428 \text{ m}^3/\text{kg}$
Temperature	$T(\mathbf{K}) = T(^{\circ}\mathbf{C}) + 273.15$ $\Delta T(\mathbf{K}) = \Delta T(^{\circ}\mathbf{C})$	$T(R) = T(^{\circ}F) + 459.67 = 1.8T(K)$ $T(^{\circ}F) = 1.8 T(^{\circ}C) + 32$ $\Delta T(^{\circ}F) = \Delta T(R) = 1.8* \Delta T(K)$
Thermal conductivity	$1 \text{ W/m} \cdot ^{\circ}\text{C} = 1 \text{ W/m} \cdot \text{K}$	$1 \text{ W/m} \cdot ^{\circ}\text{C} = 0.57782 \text{ Btu/h} \cdot \text{ft} \cdot ^{\circ}\text{F}$
Thermal resistance	1°C/₩ 1 K/₩	$1 \text{ K/W} = 0.52750^{\circ} \text{F/h} \cdot \text{Btu}$
Velocity	1  m/s = 3.60  km/h	1  m/s = 3.2808  ft/s = 2.237  mi/h 1 mi/h = 1.46667 ft/s 1 mi/h = 1.609 km/h
Viscosity, dynamic	$1 \text{ kg/m} \cdot s = 1 \text{ N} \cdot s/m^2 - 1 \text{ Pa} \cdot s = 10 \text{ poise}$	$1 \text{ kg/m} \cdot \text{s} \approx 2419.1 \text{ lbf/ft} \cdot \text{h} \\ = 0.020886 \text{ lbf} \cdot \text{s/ft}^2 \\ \approx 5.8016 \times 10^{-6} \text{ lbf} \cdot \text{h/ft}^2$
Viscosity, kinematic	$t m^2/s = 10^4 cm^2/s$ $1 stoke = 1 cm^2/s = 10^{-4} m^2/s$	$1 \text{ m}^{2}/\text{s} = 10.764 \text{ ft}^{2}/\text{s} = 3.875 \times 10^{4} \text{ ft}^{2}/\text{h}$ $1 \text{ m}^{2}/\text{s} = 10.764 \text{ ft}^{2}/\text{s}$
Volume	$1 \text{ m}^3 = 1000 \text{ L} = 10^6 \text{ cm}^3 \text{ (cc)}$	$1 m^{3} = 6.1024 \times 10^{4} in^{3} = 35.315 ft^{3}$ = 264.17 gal (U.S.) 1 U.S. gallon = 231 in^{3} = 3.7854 L 1 fl ounce = 29.5735 cm^{3} = 0.0295735 L 1 U.S. gallon = 128 fl ounces

"Exact conversion factor between metric and English units.

<sup>4</sup>Mechanical horsepower. The electrical horsepower is taken to be exactly 746 W.





The Moody chart for the friction factor for fully developed flow in circular tubes.